

# Medium-sized cyclophanes. Part 72. Synthesis and structures of 9-methoxy(1,4)naphthaleno[3.3]metacyclophane-2,11-diones

Takehiko Yamato\*, Ryo Okabe, Shinpei Miyamoto and Minoru Miyazaki

Department of Applied Chemistry, Faculty of Science and Engineering, Saga University, Honjo-machi 1, Saga-shi, Saga 840-8502, Japan

*Syn*- and *anti*-9-Methoxy(1,4)naphthaleno[3.3]MCP-2,11-diones (**4**) are obtained by the coupling reaction of 2,6-bis[2-cyano-2-(toluenesulfonyl)ethyl]anisoles (**2**) and 1,4-bis(bromomethyl)naphthalene (**3**) in dimethylformamide (DMF) with an excess of sodium hydride.

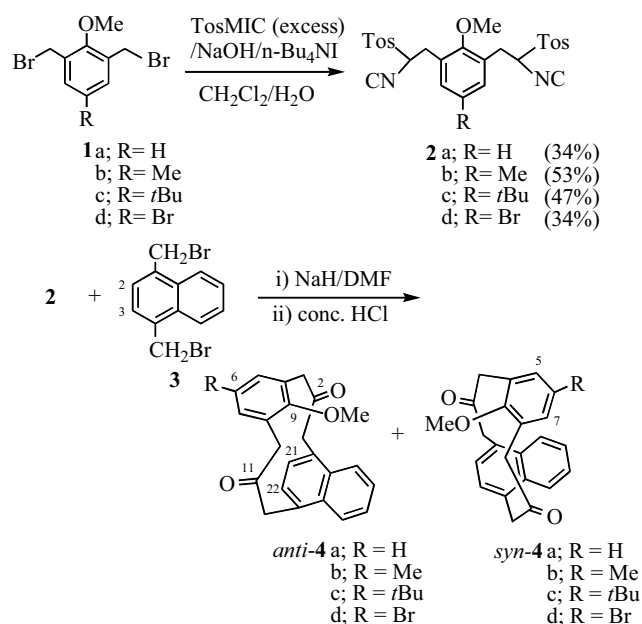
**Keywords:** metacyclophanes, cyclisation, conformation, through-space interaction

[3.3]MPCP (MPCP = metaparacyclophane) was first prepared by Shinmyozu and co-workers<sup>2</sup> using (*p*-tolylsulfonyl)methyl isocyanide (TosMIC) as the cyclisation reagent, followed by Wolff–Kishner reduction. The meta-bridged benzene ring of [3.3]MPCP has been shown to undergo conformational flipping<sup>2,3</sup> with a significantly lower energy barrier than that in [2.2]MPCP (*ca* 80 kJ mol<sup>-1</sup>).<sup>4</sup> Recently, we have reported the synthesis of 9-substituted [3.3]MPCP-2,11-diones and conversion to the corresponding [3.3]MPCPs by Wolff–Kishner reduction.<sup>5</sup> The different orientation for the acetylation was observed depending on the substituent at C (9) position.

On the other hand, we reported the preparation of 2,11-dithia(1,4)naphthaleno[3.3] MCPs (MCP = metacyclophane) and an internal substituent such as Me or OMe group is sufficient to allow the isolation of a discrete *syn* or *anti* isomer.<sup>6</sup> Thus, there is substantial interest that employing a naphthalene ring instead of a benzene ring of the para-bridged ring will provide good information about the  $\pi$ – $\pi$ -interaction between the two stacking aromatic rings. Furthermore, the conformations of 9-substituted [3.3]MCPs having a naphthalene skeleton are so far not known in spite of the formation of two conformers, *i.e.* *syn*- and *anti*-conformers, being possible like 2,11-dithia(1,4)-naphthaleno[3.3]MCPs. In this paper, we report on the synthesis and the structures of *syn*- and *anti*-9-methoxy(1,4)naphthaleno[3.3]MCP-2,11-diones (**4**).

## Results and discussion

Vögtle reported<sup>7</sup> the preparation of [3<sub>n</sub>]MCP-triones using (*p*-tolylsulfonyl)methyl isocyanide (TosMIC)<sup>8</sup> as the cyclisation reagent, which was applied in a new cyclisation procedure without phase-transfer conditions.<sup>9</sup> This strategy can be employed for the preparation of (1,4)naphthaleno[3.3]MCP-2,11-diones containing two aryl rings. In fact, we have selected the stepwise cyclisation of TosMIC adduct **2** with 1,4-bis(bromomethyl)naphthalene (**3**) to prepare the desired cyclic diketones **4** as shown in Scheme 1. The starting compound, 1,4-bis(bromomethyl)naphthalene (**3**) was prepared from 1,4-dimethylnaphthalene according to the reported procedure.<sup>10</sup> The preparations of the 2,6-bis(bromomethyl)anisoles (**1a–d**)



**Scheme 1**

have already been described in earlier paper.<sup>11</sup> TosMIC adducts **2a**, **2b** and **2d** were obtained in 34–53% yield by the reaction of **1a**, **1b** and **1d** with TosMIC as a mixture of two isomers, *i.e.* *meso* and *dl*. However, the attempted separation of these isomers of **2** pure failed. The preparation of the TosMIC adduct **2c** of 2,6-bis(bromomethyl)-4-*tert*-butylanisole (**1c**) has already been described in earlier paper.<sup>12</sup> *syn*-9-Methoxy(1,4)naphthaleno[3.3]MCP-2,11-dione (**4a**) was obtained in 12% yield by the coupling reaction of 2,6-bis[2-cyano-2-(toluenesulfonyl)ethyl]anisole (**2a**) and **3** in dimethylformamide (DMF) with an excess of sodium hydride according to the reported procedure.<sup>12</sup> Similarly, in the case of 2,6-bis[2-cyano-2-(toluenesulfonyl)ethyl]-4-methylanisole (**2b**) and 4-bromoanisole (**2d**) *syn*-isomers (**4b**) and (**4d**) were predominantly obtained in 36 and 21%

**Table 1** *Anti*-to-*syn* Ratios in TosMIC cyclisation of **2** with **3**

	Substrate		Product yield/% <sup>a</sup>	Isomer distribution/% <sup>b</sup>	
	R			<i>anti</i>	<i>syn</i>
<b>2a</b>	H	<b>4a</b>	(12)	0	100
<b>2b</b>	Me	<b>4b</b>	(36)	0	100
<b>2c</b>	<i>t</i> Bu	<b>4c</b>	(45)	22 (10)	78 (35)
<b>2d</b>	Br	<b>4d</b>	(21)	0	100

<sup>a</sup>Isolated yields. <sup>b</sup>*anti*-to-*syn* Ratios determined by <sup>1</sup>H NMR spectroscopy at 20°C.

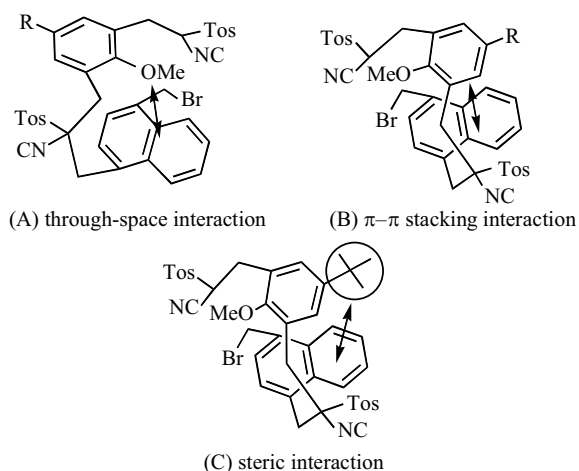
\* Correspondent. E-mail: yamatot@cc.saga-u.ac.jp

yields, respectively. No *anti*-product was obtained under the conditions used. In contrast, similar reaction of 2,6-bis[2-cyano-2-(toluenesulfonyl)ethyl]-4-*tert*-butylanisole (**2c**) with **3** afforded a mixture of *syn*-(*syn*-**4c**) and *anti*-9-methoxy(1,4)naphthaleno[3.3]MCP-2,11-dione (*anti*-**4c**) in a ratio of 78: 22 in 45% yield. Thus, depending on the substituents at position 4 on the 2,6-bis[2-cyano-2-(toluenesulfonyl)ethyl]anisoles **2**, different yields of *anti*-**4** and *syn*-**4** were achieved.

The structures of **4** have been elucidated by elemental analyses and spectral data. For instance, the mass spectral data for *anti*-**4c** ( $M^+ = 400$ ) strongly supports cyclic dimeric structure. The IR spectrum of *anti*-**4c** shows the absorption of the carbonyl stretching vibration around 1688  $\text{cm}^{-1}$ . The  $^1\text{H}$  NMR spectrum (in  $\text{CDCl}_3$ ) of *anti*-**4c** exhibits two sets of doublets at  $\delta$  3.16, 3.81 ppm ( $J = 11.0$  Hz) and 3.52, 3.81 ppm ( $J = 14.0$  Hz) for the  $\text{ArCH}_2\text{COCH}_2\text{Ar}$  methylene protons and a singlet for the methoxy protons at an upfield shift  $\delta$  2.77 ppm from 4-*tert*-butyl-2,6-dimethylanisole ( $\delta$  3.83 ppm) due to the ring current of the opposing aromatic ring.<sup>4</sup> The same upfield shift of the inner naphthalene protons ( $\text{H}_{21}$ ,  $\text{H}_{22}$ ) was observed at  $\delta_{\text{MCP}}^{21,22}$  6.36 ppm in *anti*-**4c** [ $\Delta\delta = 0.81$  ppm from 1,4-dimethylnaphthalene,  $\delta_{\text{BMX}}^{2,3}$  7.17 ppm] due to the ring current effect by the opposing benzene ring. These observations strongly suggest that compound *anti*-**4c** adopts the *anti*-conformation.

In contrast, the methoxy protons of *syn*-**4c** are observed at  $\delta$  3.33 ppm. Further, the benzene protons ( $\text{H}_5$ ,  $\text{H}_7$ ) can clearly be seen to be shielded at  $\delta_{\text{MCP}}^{5,7}$  6.45 ppm by the adjacent naphthalene ring, a common consequence of face-to-face aryl rings.<sup>4</sup> Also the *tert*-butyl proton was observed at higher field,  $\delta$  0.86 ppm compared to that of the *anti*-**4c** at  $\delta$  1.29 ppm due to the strong shielding effect of the naphthalene ring. These observations strongly suggest that compound *syn*-**4c** adopts *syn*-conformation. Similarly, the assignments of structures for other *syn* conformers *syn*-**4a**, *syn*-**4b** and *syn*-**4d** were readily apparent from their  $^1\text{H}$  NMR spectra.

The 9-methoxy analogues are exclusively formed as the *syn*-conformers except the *tert*-butyl group. These findings suggest that the through-space interaction between the non-bonding electron pairs of the oxygen atom of the methoxy group and the opposite naphthalene  $\pi$ -electrons of the *anti*-conformer may disfavor the formation of the latter (Fig. 1A). The exclusive formation of *syn*-conformer might be also governed by  $\pi$ - $\pi$ -stacking charge-transfer-type interactions<sup>13</sup> between the substituted benzene ring and naphthalene ring as shown in Fig. 1B. In the case of the 6-*tert*-butyl analogue the formation of *anti*-[3.3]MCP-2,11-dione *anti*-**4c** was observed



**Fig. 1** Reaction intermediate for the cyclisation to form 9-methoxy(1,4)naphthaleno[3.3]MCP-2, 11-diones (**4**).

(*syn*-to-*anti* ratio; 78: 22). This result might be attributed to the bulkiness of the *tert*-butyl group which would inhibit the formation of *syn*-**4c** (Fig. 1C).

In conclusion, the cyclisation reaction of 2,6-bis[2-cyano-2-(toluenesulfonyl)ethyl]anisoles (**2**) and 1,4-bis(bromomethyl)naphthalene (**3**) in DMF with an excess of sodium hydride exclusively afforded *syn*-(1,4)naphthaleno[3.3]MCP-2,11-diones **4**. The effect of the bulkiness of the 4-substituents of **2** such as *tert*-butyl group on the ratio of *syn*-to-*anti* conformers was observed. Further studies on the chemical properties of the two conformers *syn*- and *anti*-**4** are now in progress.

## Experiment

All melting points are uncorrected.  $^1\text{H}$  NMR spectra were recorded at 300 MHz on a Nippon Denshi JEOL FT-300 NMR spectrometer in deuteriochloroform with  $\text{Me}_4\text{Si}$  as an internal reference. IR spectra were measured as KBr pellets on a Nippon Denshi JIR-AQ20M spectrometer. Mass spectra were obtained on a Nippon Denshi JMS-HX110A Ultrahigh Performance Mass Spectrometer at 75 eV using a direct-inlet system. Elemental analyses were performed by Yanaco MT-5.

### Materials

2,6-Bis(bromomethyl)anisoles (**1a–1d**) and 2,6-bis[2-cyano-2-(toluenesulfonyl)ethyl]-4-*tert*-butylanisole (**2c**) were prepared according to the literature.<sup>11,12</sup>

### Preparation of the TosMIC adduct **2**. Typical procedure

To a mixture of 20% aqueous NaOH (25  $\text{cm}^3$ ) and  $\text{CH}_2\text{Cl}_2$  (30  $\text{cm}^3$ ) was added *n*- $\text{Bu}_4\text{NI}$  (440 mg, 1.2 mmol) followed by a solution of TosMIC (4.45 g, 25 mmol) in  $\text{CH}_2\text{Cl}_2$  (30  $\text{cm}^3$ ). After the reaction mixture was stirred at room temperature for 30 min, a solution of 2,6-bis(bromomethyl)anisole (**1a**) (3.0 g, 8 mmol) in  $\text{CH}_2\text{Cl}_2$  (30  $\text{cm}^3$ ) was added. The reaction mixture was stirred at room temperature for 2 h, quenched with water (50  $\text{cm}^3$ ), and was extracted with  $\text{CH}_2\text{Cl}_2$  (50  $\text{cm}^3 \times 3$ ). It was washed with water (50  $\text{cm}^3$ ), dried with  $\text{Na}_2\text{SO}_4$ , and concentrated *in vacuo* to leave a residue. To this residue methanol (50  $\text{cm}^3$ ) was added and left overnight in the refrigerator to give 2,6-bis[2-cyano-2-(toluenesulfonyl)ethyl]anisole (**2a**) (1.64 g, 34%) as pale brown prisms; m.p. 104–106°C;  $\nu_{\text{max}}(\text{KBr})/\text{cm}^{-1}$ : 2136 (CN);  $\delta_{\text{H}}(\text{CDCl}_3)$ : 2.49 (6 H, s, Me), 3.05 (2 H, m,  $\text{CH}_2$ ), 3.67 (2 H, dd,  $J = 2.9$ , 2.8,  $\text{CH}_2$ ), 3.79 (3 H, s, OMe), 4.75 (2 H, dd,  $J = 3.1$ , 3.1, CH), 7.09 (1 H, d,  $J = 7.0$ , Ar-H), 7.19 (1 H, d,  $J = 7.0$ , Ar-H), 7.24 (1 H, t,  $J = 6.2$ , Ar-H), 7.44 (4 H, d,  $J = 7.9$ , Ar-H), 7.90 (4 H, d,  $J = 8.3$ , Ar-H);  $m/z$ : 522 ( $M^+$ ). Anal. calcd. for  $\text{C}_{27}\text{H}_{26}\text{N}_2\text{O}_5\text{S}_2$  (522.6): C, 62.05; H, 5.01; N, 5.36. Found: C, 62.08; H, 5.02; N, 5.21.

Compounds **2b** and **2d** were similarly prepared in 53 and 34% yields as shown in Scheme 1.

2,6-Bis[2-cyano-2-(toluenesulfonyl)ethyl]-4-methylanisole (**2b**): obtained as pale brown powder (methanol), m.p. 151–153°C (dec.);  $\nu_{\text{max}}(\text{KBr})/\text{cm}^{-1}$ : 2133 (CN);  $\delta_{\text{H}}(\text{CDCl}_3)$ : 2.27 (3 H, s, Me), 2.49 (6 H, s, Me), 2.89 (2 H, dd,  $J = 11.7$ , 13.7,  $\text{CH}_2$ ), 3.61 (2 H, dd,  $J = 2.9$ , 3.1,  $\text{CH}_2$ ), 3.79 (3 H, s, OMe), 4.75, 4.83 (2 H, dd,  $J = 3.1$ , 3.1, CH), 7.00 (2 H, s, Ar-H), 7.43 (4 H, d,  $J = 7.9$ , Ar-H), 7.90 (4 H, d,  $J = 8.4$ , Ar-H);  $m/z$ : 536 ( $M^+$ ). Anal. calcd. for  $\text{C}_{28}\text{H}_{28}\text{O}_5\text{N}_2\text{S}_2$  (536.67): C, 62.67; H, 5.26; N, 5.22. Found: C, 62.48; H, 5.24; N, 5.38.

2,6-Bis[2-cyano-2-(toluenesulfonyl)ethyl]-4-bromoanisole (**2d**): obtained as pale brown prisms; m.p. 104–106°C;  $\nu_{\text{max}}(\text{KBr})/\text{cm}^{-1}$ : 2132 (CN);  $\delta_{\text{H}}(\text{CDCl}_3)$ : 2.49, 2.50 (6 H, each s, Me), 2.89, 3.04 (2 H, each dd,  $J = 11.2$ , 13.9,  $\text{CH}_2$ ), 3.66, 3.60, 3.67 (2 H, each dd,  $J = 2.9$ , 13.9,  $\text{CH}_2$ ), 3.79, 3.83 (3 H, each s, OMe), 4.73, 4.83 (2 H, each dd,  $J = 2.9$ , 11.2, CH), 7.26, 7.34 (each s, 2 H, Ar-H), 7.44, 7.46 (4 H, d,  $J = 8.3$ , Ar-H), 7.87, 7.93 (4 H, d,  $J = 8.3$ , Ar-H);  $m/z$ : 600, 602 ( $M^+$ ). Anal. calcd. for  $\text{C}_{27}\text{H}_{25}\text{BrN}_2\text{O}_5\text{S}_2$  (601.5): C, 53.91; H, 4.19; N, 4.66. Found: C, 53.69; H, 4.20; N, 4.66.

Cyclisation of TosMIC adduct **2c** and 1,4-bis(bromomethyl)naphthalene (**3**): To a suspension of NaH (2.1 g, 51 mmol) in DMF (150  $\text{cm}^3$ ) a solution of **2c** (4.0 g, 6.9 mmol) and 1,4-bis(bromomethyl)naphthalene (**3**) (2.23 g, 6.9 mmol) in DMF (35  $\text{cm}^3$ ) was added dropwise over a period of 6 h. After the suspension was stirred for an additional 5 h at room temperature, it was quenched with ice water (300  $\text{cm}^3$ ). The reaction mixture was extracted with  $\text{CH}_2\text{Cl}_2$  (100  $\text{cm}^3 \times 3$ ), washed with water (200  $\text{cm}^3$ ), dried with  $\text{Na}_2\text{SO}_4$ , and concentrated *in vacuo* to 30  $\text{cm}^3$ . Concentrated HCl (15  $\text{cm}^3$ ) was added, and the solution was stirred for 15 min. The organic layer was again extracted with  $\text{CH}_2\text{Cl}_2$  (100  $\text{cm}^3 \times 3$ ), washed with water

(100 cm<sup>3</sup> × 2), dried with Na<sub>2</sub>SO<sub>4</sub>, and concentrated and condensed under reduced pressure. The residue was chromatographed on silica gel using benzene and benzene-CHCl<sub>3</sub> (1: 1) as eluents to give crude *syn-4c* (1.02 g, 35%) and *anti-4c* (290 mg, 10%) as a colourless solid, respectively. Recrystallisation from hexane afforded *syn-4c* (830 mg, 30%) and *anti-4c* (230 mg, 8%) as a colourless prisms.

*Syn-6-tert-butyl-9-methoxy(1,4)naphthaleno[3.3]metacyclophane-2,11-dione (syn-4c)*: Obtained as *prisms* (hexane); m.p. 198–201°C;  $\nu_{\max}$ (KBr)/cm<sup>-1</sup>: 1699 (C=O);  $\delta_{\text{H}}$ (CDCl<sub>3</sub>): 0.86 (9 H, s, *t*Bu), 3.33 (3 H, s, *OMe*), 3.12 (2 H, d, *J* = 11.0, CH<sub>2</sub>), 3.78 (2 H, d, *J* = 14.0, CH<sub>2</sub>), 4.04 (2 H, d, *J* = 11.0, CH<sub>2</sub>), 4.30 (2 H, d, *J* = 14.0, CH<sub>2</sub>), 6.45 (2 H, s, Ar-*H*<sub>5,7</sub>), 7.20 (2 H, dd, *J* = 6.5, 3.4, Ar-*H*<sub>16,17</sub>), 7.43 (2 H, s, Ar-*H*<sub>21,22</sub>), 7.58 (2 H, dd, *J* = 6.5, 3.4, Ar-*H*<sub>15,18</sub>); *m/z*: 400 (M<sup>+</sup>). Anal. calcd. for C<sub>27</sub>H<sub>28</sub>O<sub>2</sub> (400.52): C, 80.97; H, 7.05. Found: C, 81.26; H, 7.16.

*Anti-6-tert-butyl-9-methoxy(1,4)naphthaleno[3.3]metacyclophane-2,11-dione (anti-4c)*: Obtained as *prisms* (hexane); m.p. 175–178°C;  $\nu_{\max}$ (KBr)/cm<sup>-1</sup>: 1688 (C=O);  $\delta_{\text{H}}$ (CDCl<sub>3</sub>): 1.29 (9 H, s, *t*Bu), 2.77 (3 H, s, *OMe*), 3.16 (2 H, d, *J* = 11.0, CH<sub>2</sub>), 3.52 (2 H, d, *J* = 14.0, CH<sub>2</sub>), 3.81 (2 H, d, *J* = 11.0, CH<sub>2</sub>), 4.45 (2 H, d, *J* = 14.0, CH<sub>2</sub>), 6.36 (2 H, s, Ar-*H*<sub>21,22</sub>), 6.88 (2 H, s, Ar-*H*<sub>5,7</sub>), 7.57 (2 H, dd, *J* = 6.5, 3.4, Ar-*H*<sub>16,17</sub>), 8.07 (2 H, dd, *J* = 6.5, 3.4, Ar-*H*<sub>15,18</sub>); *m/z*: 400 (M<sup>+</sup>). Anal. calcd. for C<sub>27</sub>H<sub>28</sub>O<sub>2</sub> (400.52): C, 80.97; H, 7.05. Found: C, 80.68; H, 6.81.

Compounds *syn-4a*, *syn-4b* and *syn-4d* were similarly prepared in 12, 36 and 21% yields as shown in Table 1.

*Syn-9-methoxy(1,4)naphthaleno[3.3]metacyclophane-2,11-dione (syn-4a)*: Obtained as *prisms* (hexane); m.p. 250–253°C;  $\nu_{\max}$ (KBr)/cm<sup>-1</sup>: 1694 (C=O);  $\delta_{\text{H}}$ (CDCl<sub>3</sub>): 3.14 (2 H, d, *J* = 11.6, CH<sub>2</sub>), 3.34 (3 H, s, CH<sub>2</sub>), 3.77 (2 H, d, *J* = 14.5, CH<sub>2</sub>), 4.06 (2 H, d, *J* = 11.6, CH<sub>2</sub>), 4.30 (2 H, d, *J* = 14.5, CH<sub>2</sub>), 5.83 (1 H, t, *J* = 7.7, Ar-*H*<sub>6</sub>), 6.38 (2 H, d, *J* = 7.7, Ar-*H*<sub>5,7</sub>), 7.23 (2 H, dd, *J* = 6.5, 3.4, Ar-*H*<sub>15,18</sub>), 7.40 (2 H, s, Ar-*H*<sub>21,22</sub>), 7.58 (2 H, dd, *J* = 6.5, 3.4, Ar-*H*<sub>16,17</sub>); *m/z*: 344 (M<sup>+</sup>). Anal. calcd. for C<sub>23</sub>H<sub>20</sub>O<sub>3</sub> (344.41): C, 80.21; H, 5.85. Found: C, 80.37; H, 5.76.

*Syn-9-methoxy-6-methyl(1,4)naphthaleno[3.3]metacyclophane-2,11-dione (syn-4b)*: Obtained as *prisms* (hexane); m.p. 245–246°C;  $\nu_{\max}$ (KBr)/cm<sup>-1</sup>: 1698 (C=O);  $\delta_{\text{H}}$ (CDCl<sub>3</sub>): 1.52 (3 H, s, *Me*), 3.08 (2 H, d, *J* = 11.7, CH<sub>2</sub>), 3.32 (3 H, s, *OMe*), 3.75 (2 H, d, *J* = 14.5, CH<sub>2</sub>), 3.99 (2 H, d, *J* = 11.6, CH<sub>2</sub>), 4.28 (2 H, d, *J* = 14.5, CH<sub>2</sub>), 6.15 (2 H, s, Ar-*H*<sub>5,7</sub>), 7.27 (2 H, dd, *J* = 6.5, 3.4, Ar-*H*<sub>15,18</sub>), 7.40 (2 H, s, Ar-*H*<sub>21,22</sub>), 7.56 (2 H, dd, *J* = 6.5, 3.4, Ar-*H*<sub>16,17</sub>); *m/z*: 458

(M<sup>+</sup>). Anal. calcd. for C<sub>24</sub>H<sub>22</sub>O<sub>3</sub> (358.44): C, 80.42; H, 6.19. Found: C, 80.33; H, 6.17.

*Syn-6-bromo-9-methoxy(1,4)naphthaleno[3.3]metacyclophane-2,11-dione (syn-4d)*: Obtained as *prisms* (hexane); m.p. 289–290°C;  $\nu_{\max}$ (KBr)/cm<sup>-1</sup>: 1699 (C=O);  $\delta_{\text{H}}$ (CDCl<sub>3</sub>): 3.07 (2 H, d, *J* = 11.7, CH<sub>2</sub>), 3.33 (3 H, s, *OMe*), 3.78 (2 H, d, *J* = 14.5, CH<sub>2</sub>), 3.99 (2 H, d, *J* = 11.7, CH<sub>2</sub>), 4.33 (2 H, d, *J* = 14.5, CH<sub>2</sub>), 6.47 (2 H, s, Ar-*H*<sub>5,7</sub>), 7.39 (2 H, s, Ar-*H*<sub>15,18</sub>), 7.40 (4 H, dd, *J* = 6.5, 3.4, Ar-*H*<sub>21,22</sub>), 7.56 (2 H, dd, *J* = 6.5, 3.4, Ar-*H*<sub>16,17</sub>); *m/z*: 422, 424 (M<sup>+</sup>). Anal. calcd. for C<sub>23</sub>H<sub>19</sub>O<sub>3</sub>Br (423.31): C, 65.26; H, 4.52. Found: C, 65.24; H, 4.51.

Received 15 February 2006; accepted 12 June 2006

Paper 06/3793

## References

- 1 Medium-sized Cyclophanes. part 71: T. Yamato, T. Saisyo, T. Hironaka and S. Miyamoto, *J. Chem. Res.*, 2006, 558.
- 2 T. Shinmyozu, T. Inazu and T. Yoshino, *Mem. Fac. Sci., Kyushu Univ.*, 1985, Ser. C **15**, 79.
- 3 L. Ernst, *Progress in Nuclear Magnetic Resonance Spectroscopy*, 2000, **37**, 47.
- 4 (a) *Cyclophanes* (P.M. Keehn and S.M. Rosenfield (eds)), Academic Press: New York, vol. 1&2, 1983; (b) F. Vögtle, *Cyclophane-Chemistry*, Wiley, Chichester, 1993.
- 5 T. Yamato, K. Noda and K. Tanaka, *J. Chem. Res. (S)*, 2002, 63.
- 6 T. Yamato, K. Noda, K. Tokuhisa and M. Tashiro, *J. Chem. Res. (S)*, 1994, 210; (M), 1152.
- 7 (a) J. Breitenbach and F. Vögtle, *Synthesis*, **1992**, 41; (b) J. Breitenbach, F. Ott and F. Vögtle, *Angew. Chem.*, 1992, **104**, 360; *Angew. Chem. Int. Ed. Engl.*, 1992, **31**, 307; (c) F. Ott, J. Breitenbach, M. Nieger and F. Vögtle, *Chem. Ber.*, 1993, **126**, 97.
- 8 O. Possel and A.M. van Leusen, *Tetrahedron Lett.*, **1977**, 4229; (b) D. van Leusen, A.M. van Leusen, *Tetrahedron Lett.*, **1977**, 4233.
- 9 (a) K. Kobiro, M. Takashi, N. Nishikawa, K. Kikuchi, Y. Tobe and Y. Odaira, *Tetrahedron Lett.*, 1987, **28**, 3825; (b) K. Sako, T. Meno, H. Takemura, T. Shinmyozu and T. Inazu, *Chem. Ber.*, 1990, **123**, 630; (c) K. Sako, T. Shinmyozu, H. Takemura, M. Suenaga and T. Inazu, *J. Org. Chem.*, 1992, **57**, 6536.
- 10 M.W. Haenel, *Chem. Ber.*, 1982, **115**, 1425.
- 11 T. Yamato, T. Furukawa, K. Tanaka, T. Ishi-i and M. Tashiro, *Can. J. Chem.*, 2003, **81**, 244.
- 12 T. Yamato, L.K. Doamekpor, K. Koizumi, K. Kishi, M. Haraguchi and M. Tashiro, *Liebigs Ann.*, 1995, 1259.
- 13 M. Nishio and M. Horita, *Tetrahedron*, 1989, **45**, 7201.